



# Solids, Liquids, and Gases

## Part A. Vocabulary Review

**Directions:** Write the term that matches each description below in the spaces provided. Then unscramble the letters in the boxes to reveal the mystery term.

1. \_\_\_\_\_  \_\_\_\_\_
2. \_\_\_\_\_  \_\_\_\_\_
3. \_\_\_\_\_  \_\_\_\_\_
4. \_\_\_\_\_  \_\_\_\_\_
5. \_\_\_\_\_  \_\_\_\_\_
6. \_\_\_\_\_  \_\_\_\_\_
7. \_\_\_\_\_  \_\_\_\_\_
8. \_\_\_\_\_  \_\_\_\_\_
9. \_\_\_\_\_  \_\_\_\_\_
10. \_\_\_\_\_  \_\_\_\_\_
11.  \_\_\_\_\_
12. \_\_\_\_\_  \_\_\_\_\_

1. assumes matter is made of small particles in constant motion
2. the force which supports objects in fluids
3. temperature at which a solid becomes a liquid
4. equals  $F/A$
5. form of matter found in lightning bolts, nuclear reactors, and stars
6. the reason a car's dashboard might crack when exposed to high temperatures
7. defined as the point at which a liquid's vapor pressure equals the atmospheric pressure
8. SI unit of pressure
9. energy released as a gas changes to a liquid
10. property of fluids which enables ships and balloons to float
11. pressure times area, or a push or pull
12. a fluid's resistance to flow
13. Mystery term: \_\_\_\_\_

**Chapter Review (continued)****Part B—Concept Review**

**Directions:** Match each theory, principle, or law in Column II with its description in Column I. Write the letter of the correct term in the blank at the left.

**Column I**

- \_\_\_\_\_ 1. All matter is made of small particles that are in motion.
- \_\_\_\_\_ 2. If the volume of a container of gas is decreased, the pressure on the gas will increase if the temperature does not change.
- \_\_\_\_\_ 3. The volume of a gas increases with increasing temperature provided the pressure does not change.
- \_\_\_\_\_ 4. The buoyant force on an object in a fluid is equal to the weight of the fluid the object displaces.
- \_\_\_\_\_ 5. Pressure applied to a fluid is transmitted unchanged throughout the fluid.
- \_\_\_\_\_ 6. As the velocity of a fluid increases, the pressure exerted by the fluid decreases.

**Column II**

- a. Boyle's law
- b. Bernoulli's principle
- c. Pascal's principle
- d. kinetic theory of matter
- e. Charles's law
- f. Archimedes' principle

**Directions:** Answer the following questions on the lines provided.

7. What are two properties of amorphous solids?

\_\_\_\_\_

\_\_\_\_\_

8. In terms of thermal expansion, why is water unusual?

\_\_\_\_\_

\_\_\_\_\_

9. Why does water behave in this unusual fashion?

\_\_\_\_\_

\_\_\_\_\_

\_\_\_\_\_

10. Why does water's boiling point decrease with increases in elevation?

\_\_\_\_\_

\_\_\_\_\_

\_\_\_\_\_